

Chapter 6 Review Chemical Bonding Answers

Ethanol Alcohol Formula - Structural and Chemical Formula Chapter 6 Review Chemical Bonding:
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Ethanol Alcohol Formula - Structural and Chemical Formula

Ethanol is a compound of carbon, hydrogen and oxygen elements was described by Antoine Lavoisier and its chemical formula was determined by Nicolas-Theodore de Saussure in 1808. Five years later, Archibald Scott Couper published the structural formula of ethanol.

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Ionic bonding is a type of chemical bond that involves the electrostatic attraction between oppositely charged ions, and is the primary interaction occurring in ionic compounds. The ions are atoms that have lost one or more electrons (termed cations) and atoms that have gained one or more electrons (termed anions). This transfer of electrons is termed electrovalence in contrast to covalence.

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